

The Global Climate Thermostat: Fact, fiction, and computer models Andy Ridgwell



Regulation of global climate





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From: Hönisch et al. [2012]



















Terrestrial weathering can be (approximately equally) divided into carbonate (CaCO₃) and calcium-silicate ('CaSiO₃') weathering: (1) $2CO_{2(aq)} + H_2O + CaSiO_3 \rightarrow Ca^{2+} + 2HCO_3^{-} + SiO_2$ (2) $CO_{2(aq)} + H_2O + CaCO_3 \rightarrow Ca^{2+} + 2HCO_3^{-}$

Ultimately, the (alkalinity: Ca²⁺) weathering products must be removed through carbonate precipitation and burial in marine sediments:

(3) $Ca^{2+} + 2HCO_3^{-} \rightarrow CO_{2(aq)} + H_2O + CaCO_3$

It can be seen that in (2) + (3), that the CO_2 removed (from the atmosphere) during weathering, is returned upon carbonate precipitation (and burial). In (1) + (3) (silicate weathering) CO_2 is permanently removed to the geological reservoir. This CO_2 must be balanced by mantle (/volcanic) out-gassing on the very long term.







Furthermore, the rate of silicate weathering should scale with climate. Hence the **silicate weathering feedback** is formed:

higher $pCO_2 \rightarrow$ higher temperatures (& rainfall) \rightarrow higher weathering rates \rightarrow lower pCO_2

Outline







```
! calculate carbonate alkalinity
loc ALK DIC = dum ALK &
& - loc H4BO4 - loc OH - loc HPO4 - 2.0*loc PO4 - loc H3SiO4 - loc NH3 - loc HS &
\& + loc_H + loc_HSO4 + loc_HF + loc_H3PO4
! estimate the partitioning between the aqueous carbonate species
loc zed = (\&
& (4.0*loc ALK DIC + dum DIC*dum carbconst(icc k) -
loc ALK DIC*dum carbconst(icc k))**2 + &
& 4.0*(dum carbconst(icc k) - 4.0)*loc ALK DIC**2 &
& )**0.5
            loc conc HCO3 = (dum DIC*dum carbconst(icc k) -
loc_zed) / (dum_carbconst(icc k) - 4.0)
loc conc CO3 = \&
& ( &
   loc ALK DIC*dum carbconst(icc k) - dum DIC*dum carbconst(icc k) - &
&
&
   4.0*loc ALK DIC + loc zed &
& ( &
\& / (2.0*(dum carbconst(icc k) - 4.0))
loc_conc_CO2 = dum_DIC - loc_ALK_DIC + &
& ( &
& loc ALK DIC*dum carbconst(icc k) - dum DIC*dum carbconst(icc k) - &
& 4.0*loc ALK DIC + loc zed &
& ) &
\& / (2.0*(dum carbconst(icc k) - 4.0))
loc_H1 = dum_carbconst(icc_k1)*loc_conc_CO2/loc_conc_HCO3
loc H2 = dum carbconst(icc k2)*loc conc HCO3/loc conc CO3
```



lies, damn lies, and computer models





lies, damn lies, and computer models





http://www.seao2.info/misc_usc2017.html











(1) Series of 1 Myr Earth system model experiments. CO₂ emissions from 1,000 to 20,000 PgC (GtC). Release interval: 1 yr.





(2) Fit each CO_2 decay curve with a series (4 optimal) of exponentials. Extract the fraction of CO_2 and time-scale associated with each.

(The resulting empirical model can be used in place of a mechanistic model for projecting the long-term fate of carbon release.)

 10^{6}



Cross-plot of the fraction of total CO₂ emissions to the atmosphere removed by a particular process (carbon sink), vs. the characteristic (efolding) time-scale of that process (log₁₀ scale).











 $\diamond \longrightarrow$



Response of fraction of CO₂ removed vs. the characteristic time-scale, as a function of total emissions, ranging from 1,000 PgC (dark blue) to 20,000 PgC (yellow).





Depletion of mixed layer carbonate buffer; ocean stratification and reduced surface mixing. Warming and reduced CO₂ solubility.







Ocean stratification and collapse of the AMOC (in this particular model). Threshold reached @ ~4000 PgC?















III: τ ~ 1000-10000 years

Geologic CO₂ removal via carbonate rocks and marine sediments – occurring on an increasing protracted time-scale.















Sediments spanning the Palaeocene-Eocene boundary from ODP Leg 208 (Walvis Ridge) Picture courtesy of Dani Schmidt (University of Bristol)



















With increasing total CO_2 emissions, the response time of all sinks (bar silicate weathering) lengthen, and the shorter time-scale two weaken at the expense of the ~10,000 year CaCO₃ burial process.

Elevated atmospheric pCO_2 hence becomes more persistent as the main shortterm CO_2 feedbacks weaken.

The majority of carbon removal beyond ~10,000 PgC is removed only on time-scales exceeding 10,000 years.

Lord et al. [2015a,b]









Earth system model (CO₂ and mean SST trajectories) Downscaling (SO SST and regional climate) Ice sheet model

































granite \approx		
SiO_2	=	72%
CaO	=	1.8%
MgO	_	0.7%

basalt ≈

SiO₂ = 50% ... CaO = 10% ... MgO = 10%

• • •









Dunite







Taylor et al. [2015] (Nature Climate Change)











```
Current global oil
consumption =
90,136\times10^3 barrels per
day
1.0 barrel = 159 l
= 159\times10^3 cm<sup>3</sup>
\Rightarrow oil consumption
= 5.23\times10^{15} cm<sup>3</sup> year<sup>-1</sup>
= 5.23 km<sup>3</sup> year<sup>-1</sup>
```

```
B
```

```
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⇒ oil consumption
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How many Yosemite Valleys? (equivalent volume)





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```

```
Yosemite Valley
(Wikipedia):
1,200m deep × 1,600m
across, 12.0 km long
⇒
volume = 1.2×1.6×12.0
= 23.0 km<sup>3</sup>
```

How many Yosemite Valleys? (equivalent volume)



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